

# Gizmo Chemical Equations Answer Key

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### Gizmo: Balancing Chemical Equations

1 Match: Most chemical reactions can be classified as one of four types Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction Using the Gizmo, match the descriptions below to the correct type of reaction \_\_\_ One reactant is broken down into two or more products A Synthesis

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### **Student Exploration: Chemical Equations**

equations Get the Gizmo ready: Erase the chemical formulas in each text box Introduction: In a chemical reaction, the reactants are the substances that enter into the reaction, and the products are the substances that are made in the reaction A chemical reaction is balanced if the numbers of reactant atoms match the numbers of product atoms

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### **Student Exploration: Balancing Chemical Equations**

Gizmo Warm-up In a chemical reaction, reactants interact to form products This process is summarized by a chemical equation In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction:  $H_2 + O_2 \rightarrow H_2O$  How many atoms are in a hydrogen molecule ( $H_2$ )? \_\_\_ 2 How many atoms are in an oxygen

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### Cell Energy Cycle

4 Summarize: A chemical equation shows reactants on the left side of an arrow, and products on the right, like this: reactant + reactant product + product Based on your observations, what is the chemical equation for photosynthesis?  $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$  Turn on Show chemical equation to check Were you correct? Check answers 5

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### Balancing Chemical Equations Answer Key

Balancing Chemical Equations -Answer Key Balance the equations below: 1)  $1 \text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3$  2)  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  3)  $2 \text{NaCl} + 1 \text{F}_2 \rightarrow 2 \text{NaF} + 1 \text{Cl}_2$  4)  $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$  5)  $1 \text{Pb}(\text{OH})_2 + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O} + 1 \text{PbCl}_2$  6)  $2 \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2(\text{SO}_4)_3$

### Student Exploration: Chemical Equations - PAP

Student Exploration: Chemical Equations - PAP Login to [www.explorelearning.com](http://www.explorelearning.com) to complete this assignment Define the following terms on a separate paper by 10/1: Coefficient, Conservation of matter, Molecule, Product, Reactant, Subscript Prior Knowledge Questions (Do these BEFORE using the Gizmo) 1 A candle is placed on one pan of a balance

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